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Effective Date: December 16, 2013 22nd Edition of Standard Methods for the Examination of Water and Wastewater ERRATA Despite the best efforts of those involved in the manufacture of the 22nd Edition of Standard Methods, the following errors have occurred in the text: 1. In Section 1020B.1 (p. 1-8), the second equation should be "Lower control limit = Mean

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For methods where bias is applicable (see Table 2020:I), run a laboratory-fortified blank (LFB) (2020B.2e), performance evaluation sample, or standard with a known or otherwise certifiable concentration at least four times and compare results to the limits listed in the method or those established by the laboratory. If no limit is specified, use the following procedure to establish limits:

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where: pH = log₁₀ a H⁺ and. pOH = log₁₀ a OH⁻ .. Equation 2 states that as pH increases pOH decreases correspondingly and vice versa because pK_w is constant for a given temperature. At 25°C, pH 7.0 is neutral, the activities of the hydrogen and hydroxyl ions are equal, and each corresponds to an approximate activity of 10⁻⁷ moles/L. The neutral point is temperature-dependent and is ...

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The methods in the 22nd edition (as in previous editions) are believed to be the best available, generally accepted procedures for analyzing water, wastewater, and related materials. They represent the recommendations of specialists, ratified by a large number of analysts and others of more general expertise, and as such are truly consensus standards, offering a valid and recognized basis for control and evaluation.

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herein to the standard methods described elsewhere in this publication. Other sections of Part 1000 address laboratory equipment, laboratory safety, sampling procedures, and method

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