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pogil common ion effect on solubility answers

pogil common ion effect on solubility answers. In the water treatment process, sodium carbonate salt is added to precipitate the calcium carbonate. The initial value of the reaction quotient is therefore equal to zero. Through the addition of common ions, the solubility of a compound generally decreases due to a shift in equilibrium.

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Use your solubility curve graph provided to answer the following questions. 1. What are the customary

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units of solubility on solubility curves? Yemp 2. Define solubility. ok be aya 3. According to the graph, the solubility of any substance changes as changes. 4. List the substances whose solubility decreases as temperature increases. 5.

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## Webs

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Are your answers to parts a and b consistent with the information in Model 1? yes. STOP POGIL™ Activities for High School Chemistry . Model 3 - Number of Protons and Attractive Force Force of Attraction (Newtons) 0.10 nm 2.30 x 10<sup>-8</sup> 0.10 nm X 10<sup>8</sup> 0.10 nm 6.90 x

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Interpreting Solubility Curves 5. Use the temperature dependence of solubility to identify whether the substances in Table G are gases or solids. Make two lists below, one for gases and one for solids. 6. Suggest a reason why solubility decreases with increasing temperature for gaseous solutes but increases for solid solutes. Problems 1.

Interpreting Solubility Curves

b. Is your answer in part a consistent with your answer to Question 7: The enthalpy of fusion given in 'Model 2 for each substance is a measure of the energy (in kJ) needed to melt a mole of that substance. a. Which type of solid has the most endothermic enthalpies of fusion.' b.

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